Водный раствор гидроксида бария реагирует с каждым веществом в ряду:

 $1) \ NaHCO_{3}, Zn(OH)_{2}, Na_{2}SO_{4}; \\ \hspace{0.5cm} 2) \ K_{2}SO_{4}, CuO, FeCl_{2}; \\ \hspace{0.5cm} 3) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ CO_{2}, Mn_{2}O_{7}, Cu. \\ \hspace{0.5cm} 2) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ CO_{2}, Mn_{2}O_{7}, Cu. \\ \hspace{0.5cm} 2) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ CO_{2}, Mn_{2}O_{7}, Cu. \\ \hspace{0.5cm} 2) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ CO_{2}, Mn_{2}O_{7}, Cu. \\ \hspace{0.5cm} 2) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ CO_{2}, Mn_{2}O_{7}, Cu. \\ \hspace{0.5cm} 2) \ NaNO_{3}, Mg(OH)_{2}, FeCl_{3}; \\ \hspace{0.5cm} 4) \ NaNO_{3}, Mg(OH)_{2}, Mg(OH)_{2},$